

Cavitation in lipid bilayers poses strict negative pressure stability limit in biological liquids

Matej Kanduč

Jožef Stefan Institute, Ljubljana

What limits the suction intensity of plants?



In collaboration with:

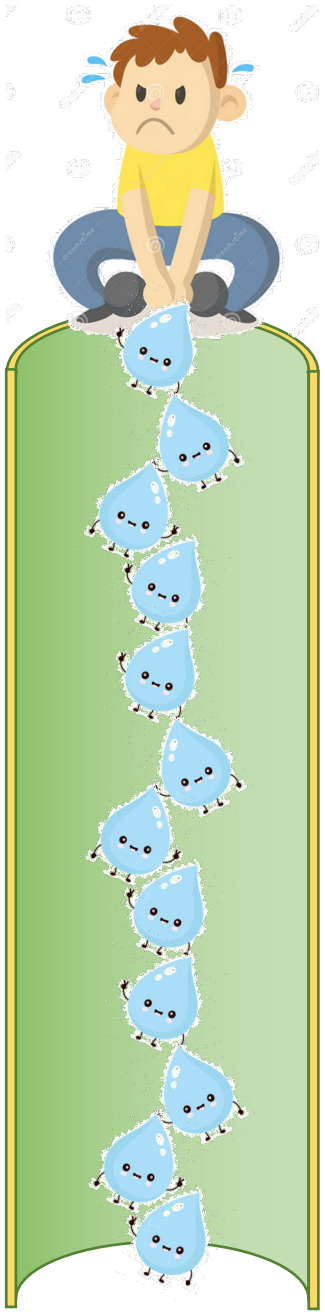
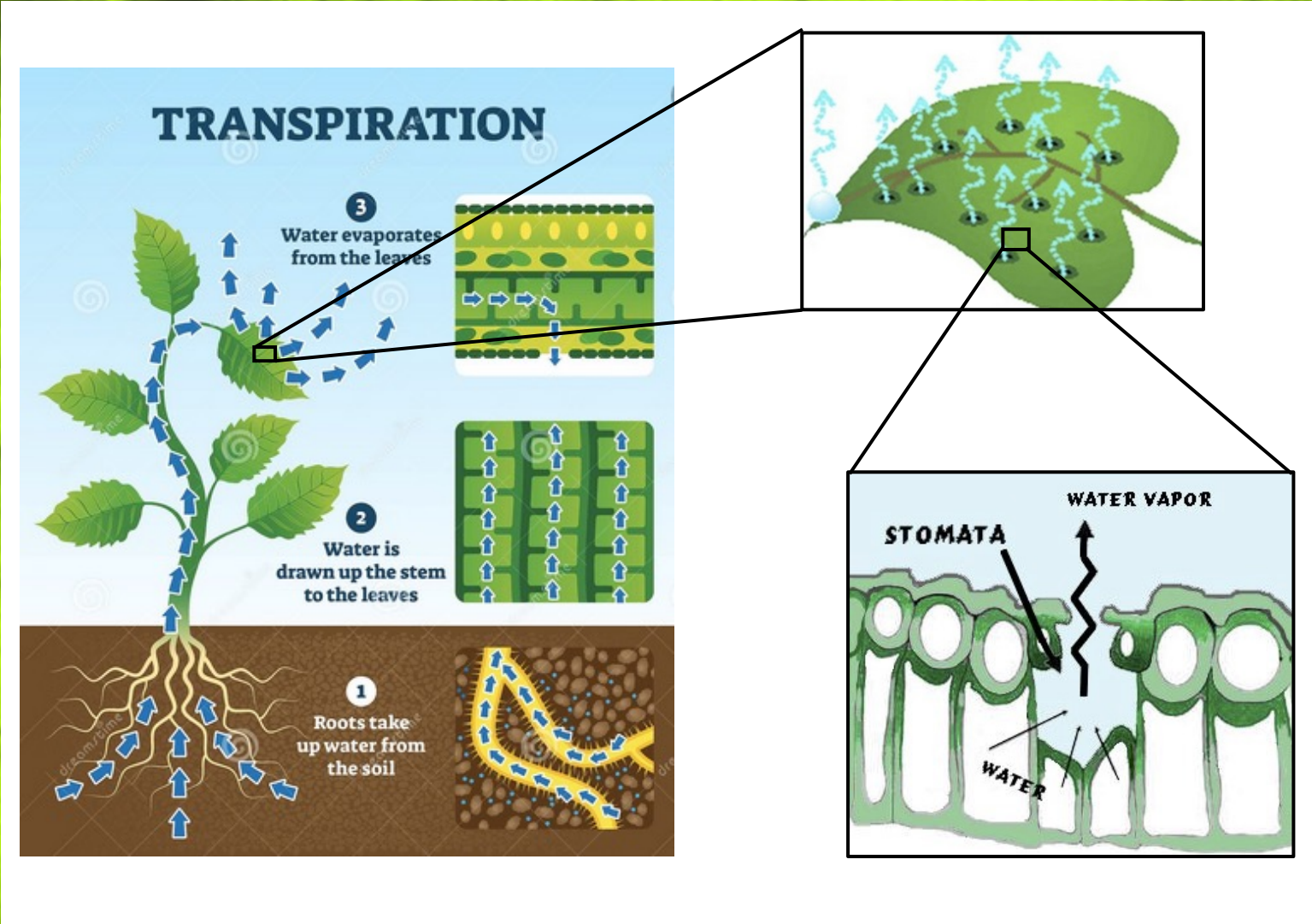
Physics

- Roland R. Netz, FU Berlin
- Emanuel Schneck, TU Darmstadt
- Philip Loche, FU Berlin

Botany

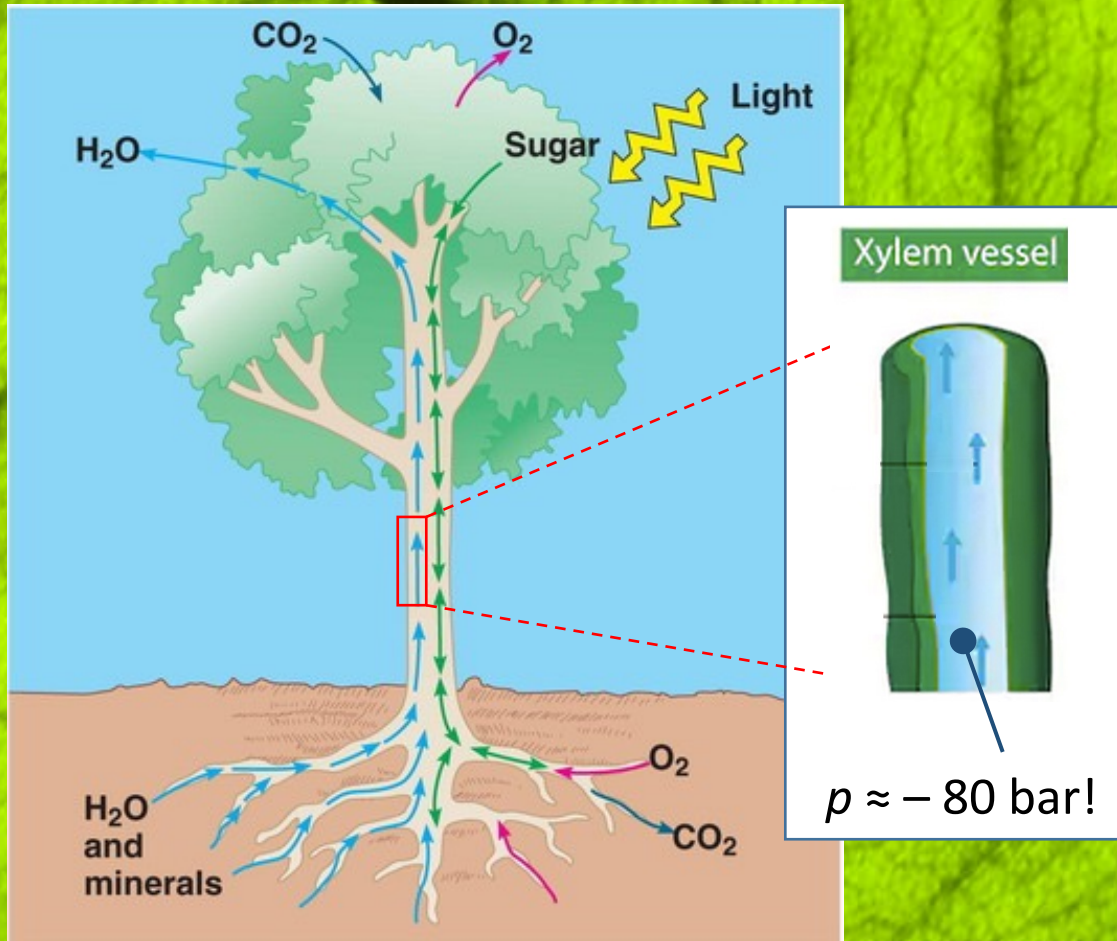
- Jochen Schenk, Cal State Fullerton
- Steven Jansen, Ulm University

Hydraulic systems in plants

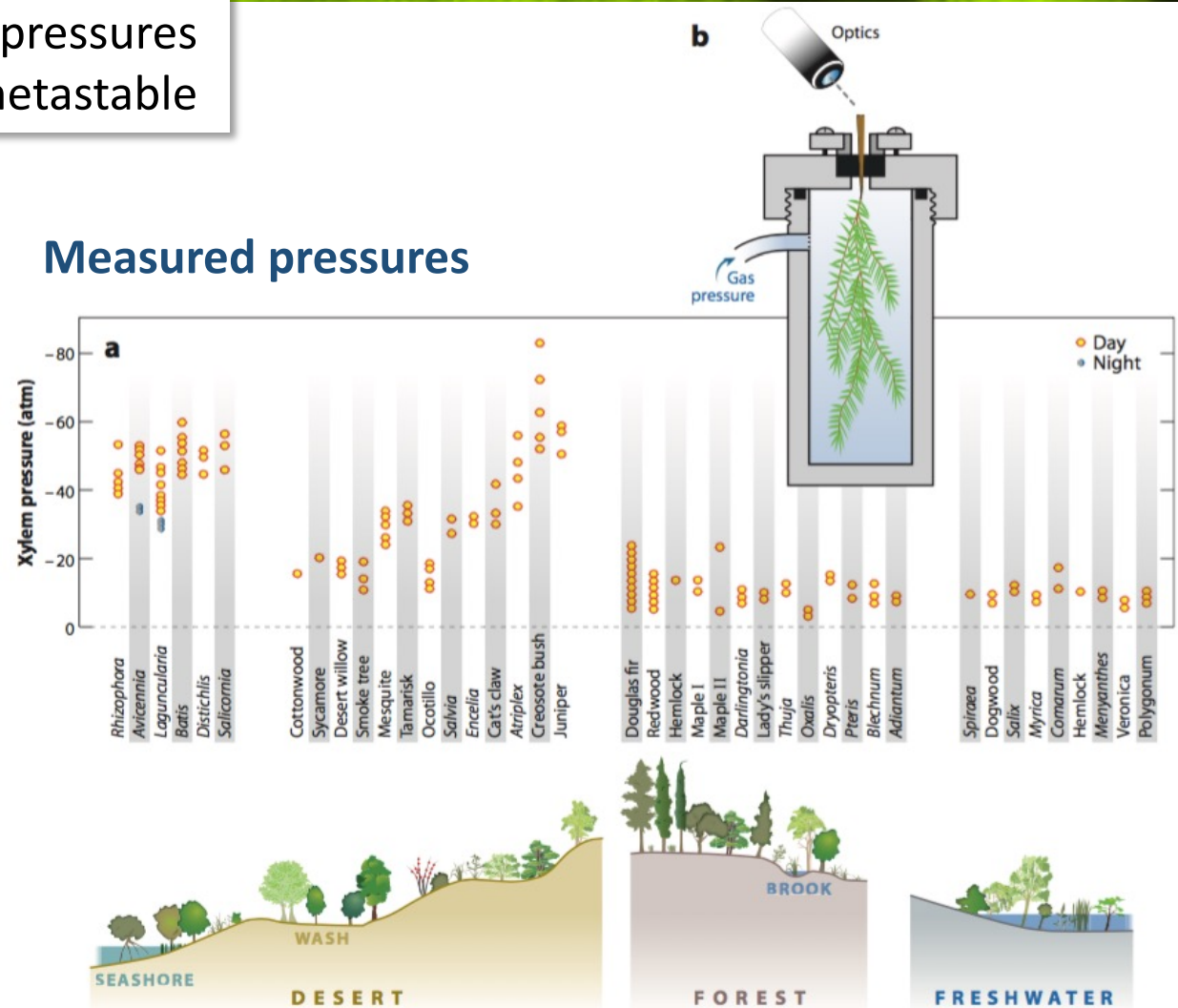


Hydraulic systems in plants

Plants transport water from the roots under negative pressures
Problem: Water and the surrounding structures are metastable



Measured pressures

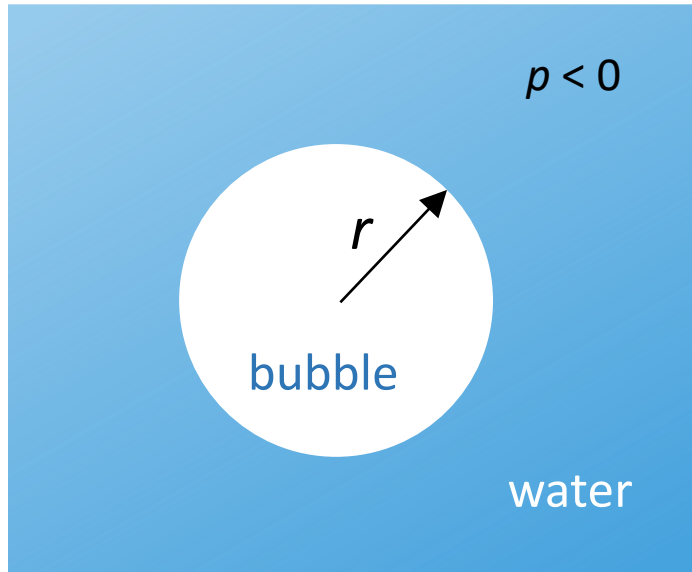


Stroock et al., Annu. Rev. Fluid Mech. 2014



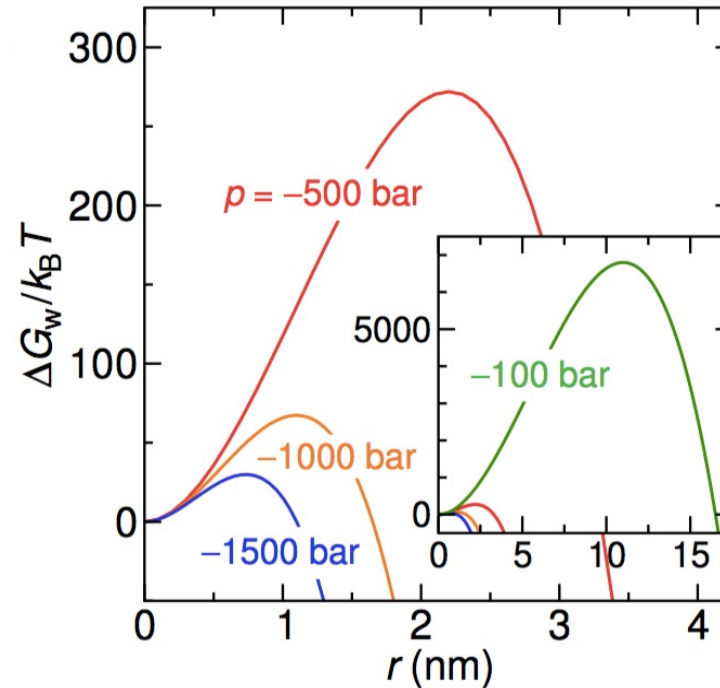
Free energy of a bubble formation

Cavitation barrier



Free energy of the bubble:

$$\Delta G_w = 4\pi r^2 \gamma + \frac{4}{3}\pi r^3 p$$



Free energy barrier:

$$\Delta G_w^* = \frac{16\pi}{3} \frac{\gamma_0^3}{p^2}$$

Transition rate theory

- Arrhenius description of the transition rate over the barrier

$$k = k_0 e^{-\beta \Delta G_w^*}$$

where

$$k_0 = \kappa_0 V$$

- cavitation rate:

$$\frac{df}{dt} = -kf$$

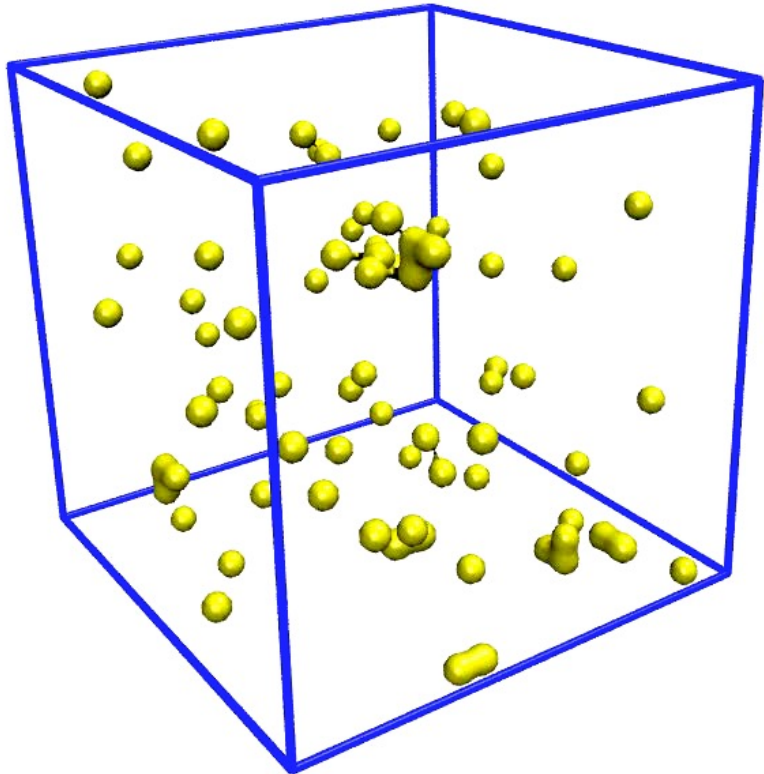
with the solution

$$f(t) = e^{-kt}$$

(survival probability)

Bubble nucleation from MD simulations

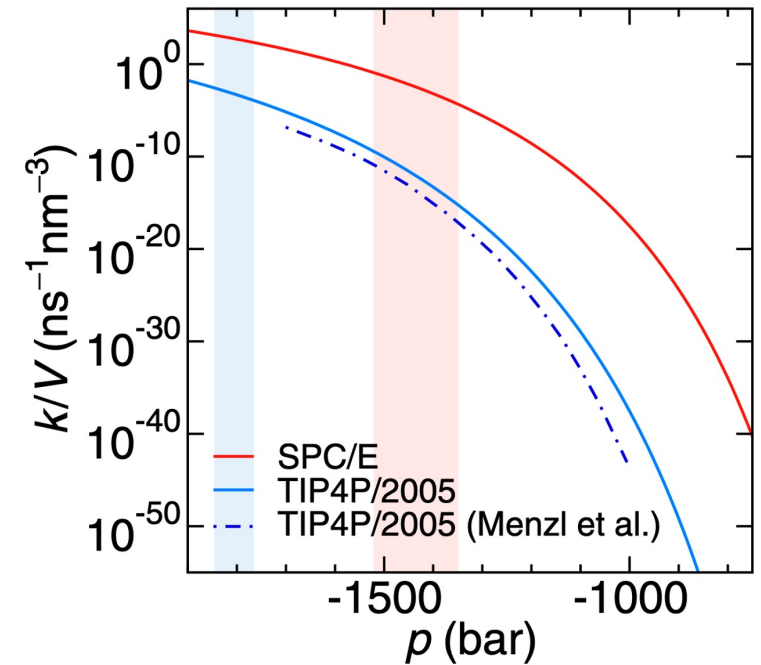
Nucleation event:
 $p = -1500$ bar



parameters: κ_0, γ

Cavitation rate density

$$J = V^{-1} \left(-\frac{df}{dt} \right)_{t=0}$$



practical example

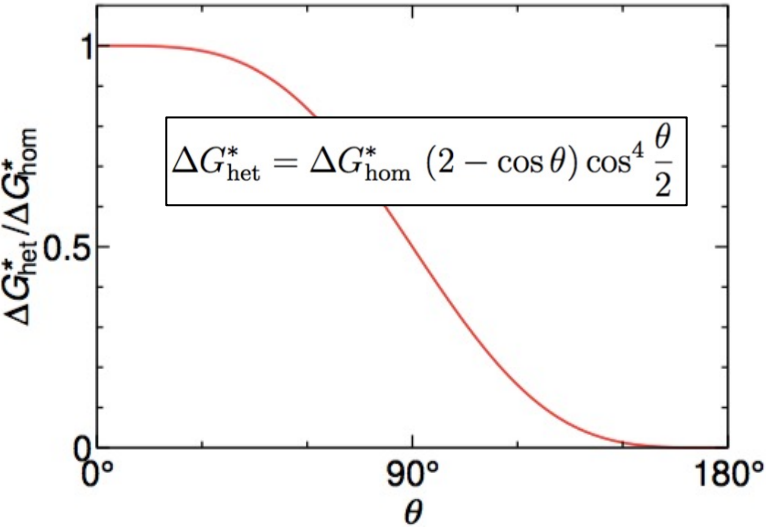
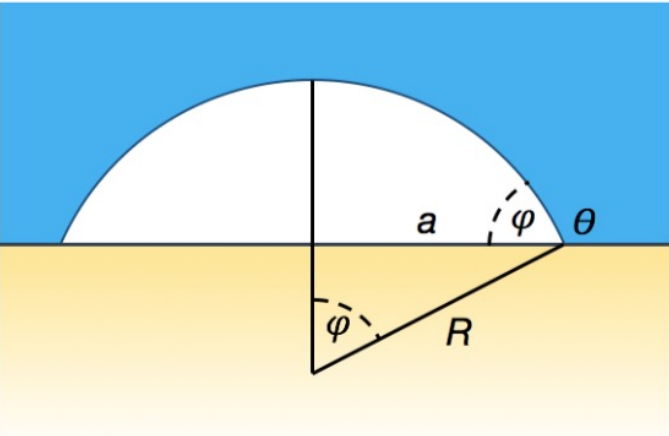
A liter (1 L) of pure water at $p = -100$ bar cavitates on average in

$$\tau_{\text{cav}} = \frac{1}{\kappa_0 V} e^{\beta \Delta G_w^*} \sim 10^{2890 \pm 100} \text{ s}$$

*note: the age of the universe is only 4×10^{17} s

Cavitation triggers (catalysts)

Heterogeneous cavitation



Pre-existing nanobubbles



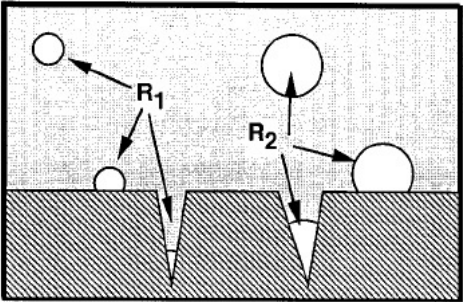
Advances in Colloid and Interface Science
80 (1999) 27–50

Bubble nucleation from gas cavities — a review

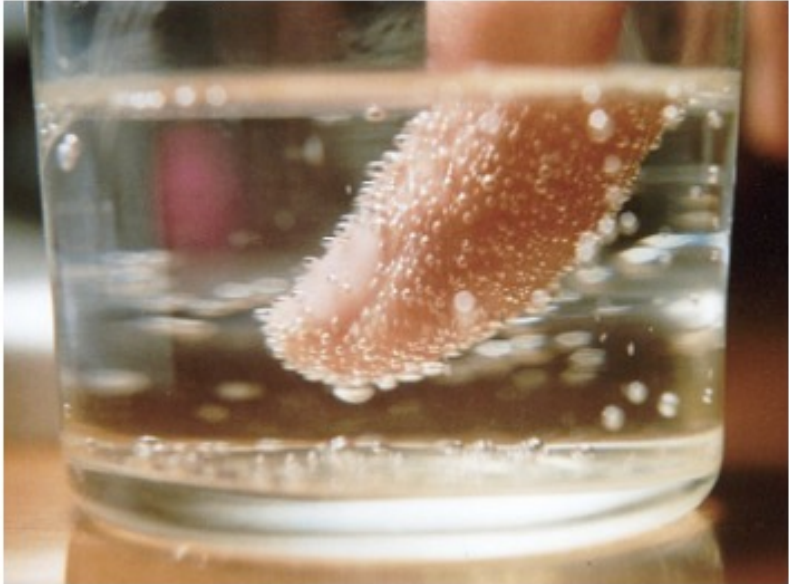
S.F. Jones, G.M. Evans, K.P. Galvin*

Type III (pseudo-classical nucleation)

Before Supersaturation
(pre-existing gas cavities)



homogeneous & heterogeneous with pre-existing gas cavities: low levels of supersaturation



Sap contains much more...

Plant Physiol. Vol. 173, 2017

Xylem Surfactants Introduce a New Element to the Cohesion-Tension Theory¹[OPEN]

H. Jochen Schenk*, Susana Espino, David M. Romo, Neda Nima, Aissa Y.T. Do, Joseph M. Michaud, Brigitte Papahadjopoulos-Sternberg, Jinlong Yang, Yi Y. Zuo, Kathy Steppe, and Steven Jansen

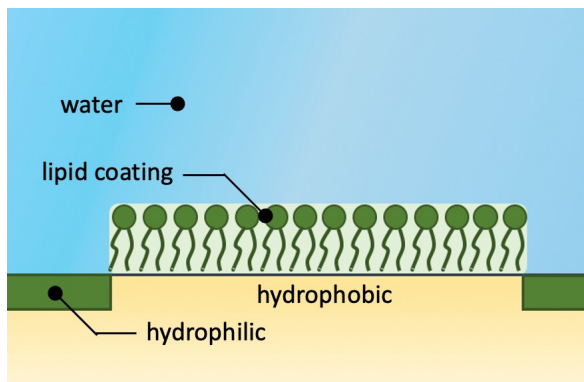
AMERICAN JOURNAL OF BOTANY 102(10): 1561–1563, 2015;

On the ascent of sap in the presence of bubbles¹

Steven Jansen^{2,4} and H. Jochen Schenk³

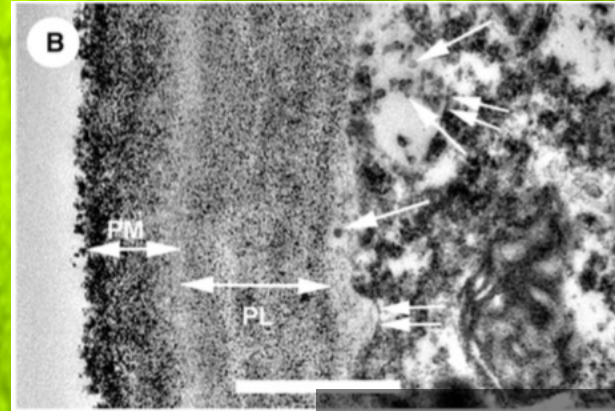
Elements of vascular water (sap):

- supersaturated with some gases
- contains surface-active molecules
- xylem contains hydrophobic surfaces

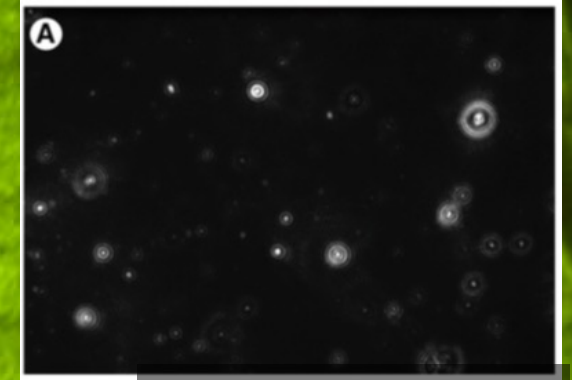


Solution?

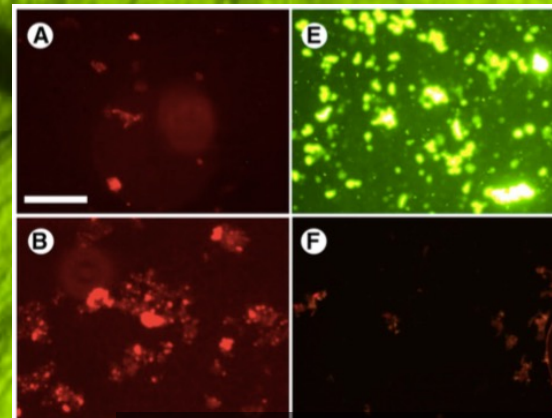
Lipids cover up hydrophobic patches



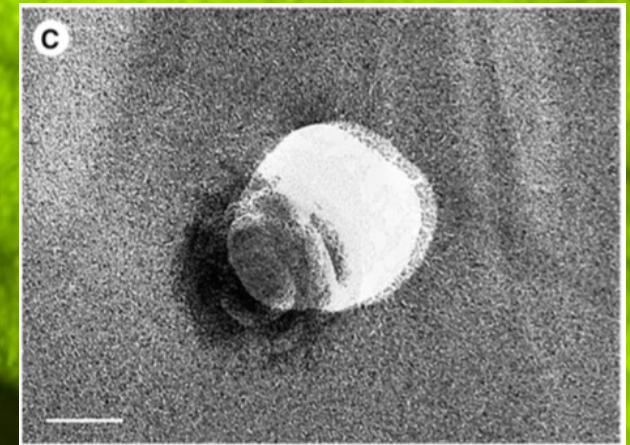
small vesicles



nanoparticles in sap

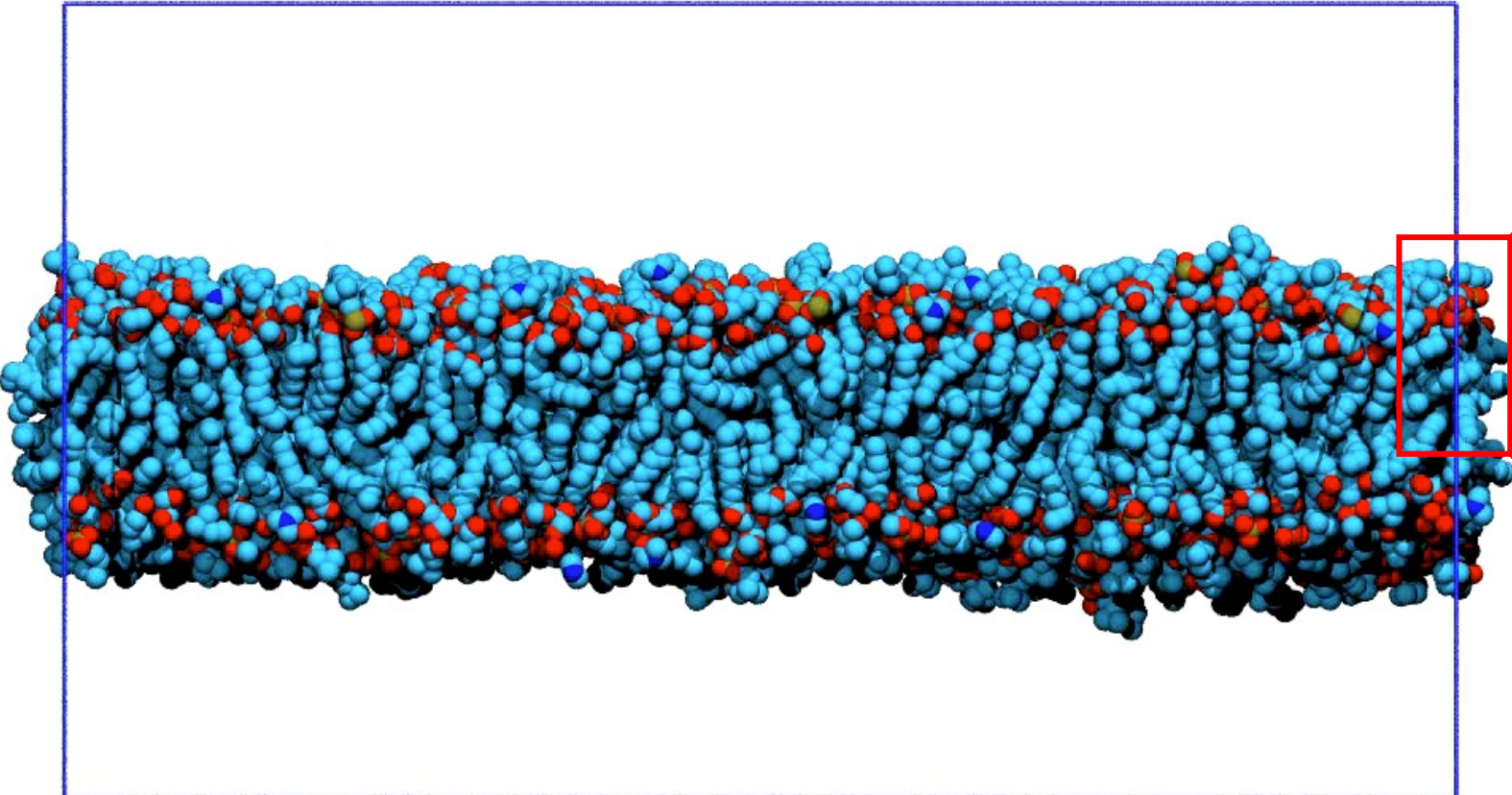


amphiphilic molecules



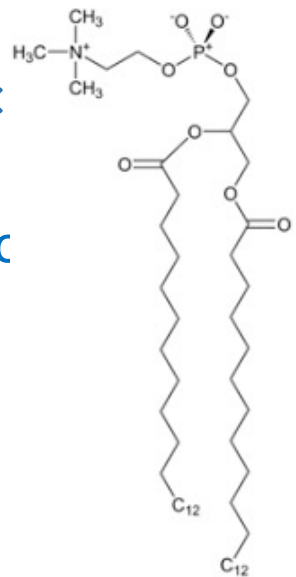
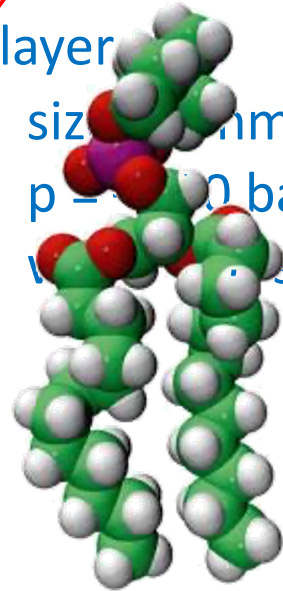
Electron micrograph of a surfactant monolayer-coated nanobubble

Cavitation of a lipid bilayer



12:0 PC (DLPC)
MD simulation of a connected bilayer

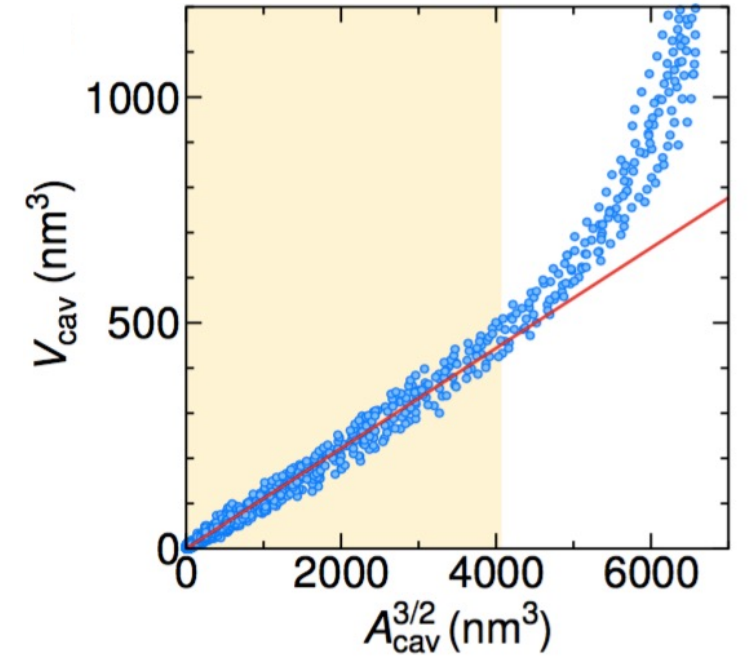
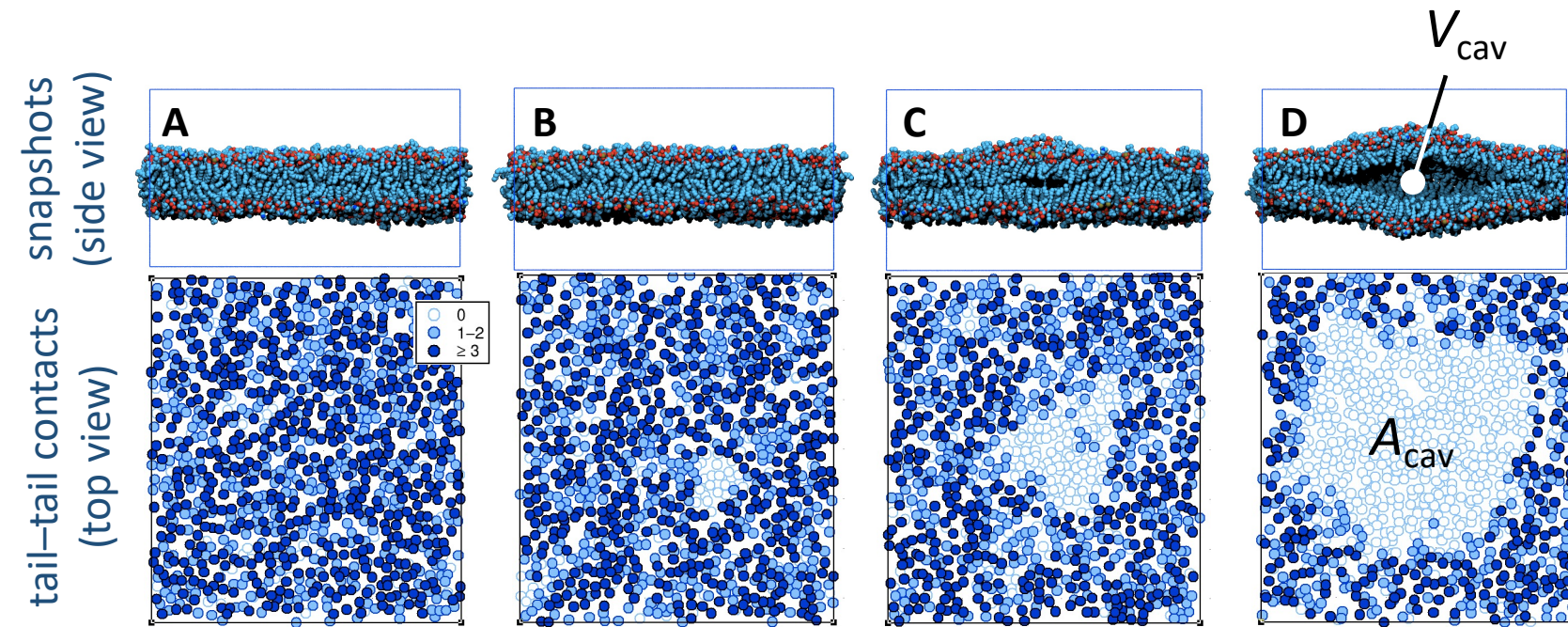
- size = 10 nm × 10 nm × 5 nm
- p = 100 bar
- volume = 1 μm³



Kanduč et al., PNAS **117**, 10733 (2020)

Lipid bilayer cavitation

Top: consecutive bilayer cross-sections (water not shown) during a cavitation event at -215 bar. The lateral surface area of the simulation box: 18 nm × 20 nm.



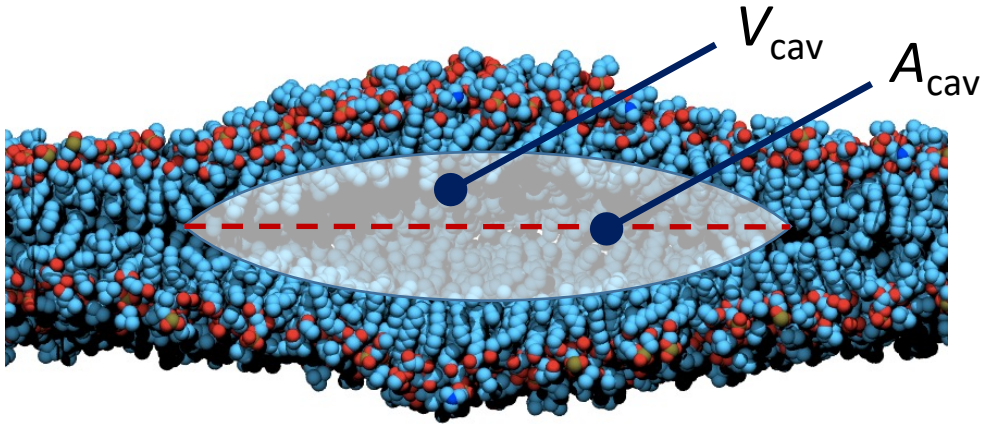
Correlation between the cavity volume and the cross section area of the cavity

$$V_{cav} = \alpha_{lip} A_{cav}^{3/2}$$

$$\alpha_{lip} = 0.11$$

Kanduč et al., PNAS **117**, 10733 (2020)

Bilayer cavitation



- Free energy of the bilayer cavity:

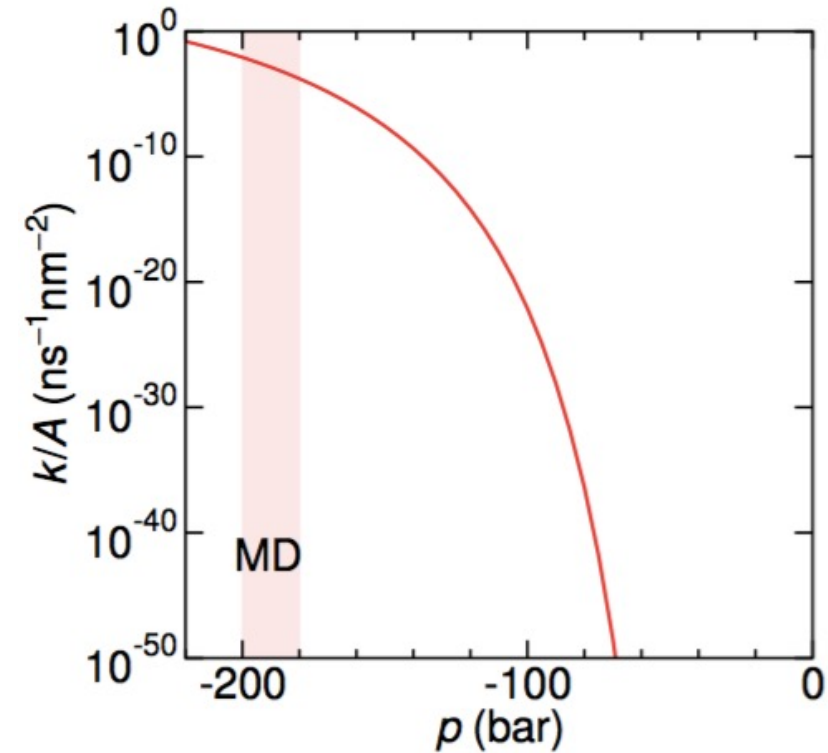
$$\Delta G_{\text{lip}} = w_{\text{lip}} A_{\text{cav}} + p V_{\text{cav}}(A_{\text{cav}}).$$

- Free energy barrier:

$$\Delta G_{\text{lip}}^* = \frac{4w_{\text{ll}}^3}{27\alpha^2 p^2}$$

- Cavitation rate:

$$k/A = \kappa_0^{\text{lip}} \exp(-\beta \Delta G_{\text{lip}}^*)$$



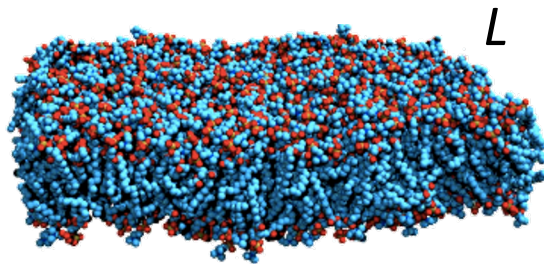
Kanduč et al., PNAS **117**, 10733 (2020)

Bilayer cavitation

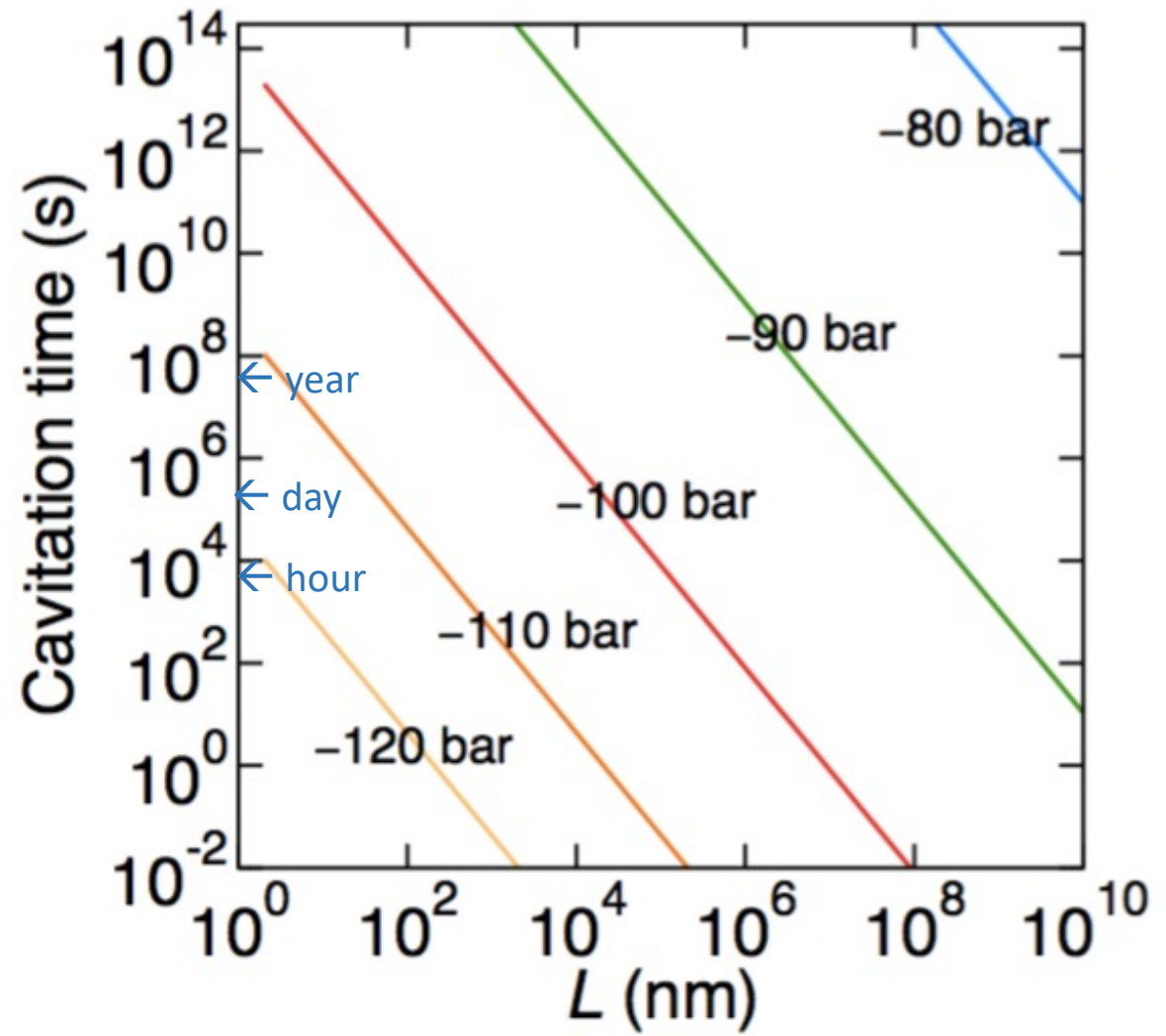
An example:

How long does it take for a bilayer to cavitate?

- constant p
- Assuming square bilayer of $L \times L$:



$$\tau_{\text{cav}} = \frac{1}{\kappa_0^{\text{lip}} A} e^{\beta \Delta G_{\text{lip}}^*}$$



Kanduč et al., PNAS **117**, 10733 (2020)

Maximal tensions

Cavitation pressure versus the mean cavitation time for different bilayer areas:

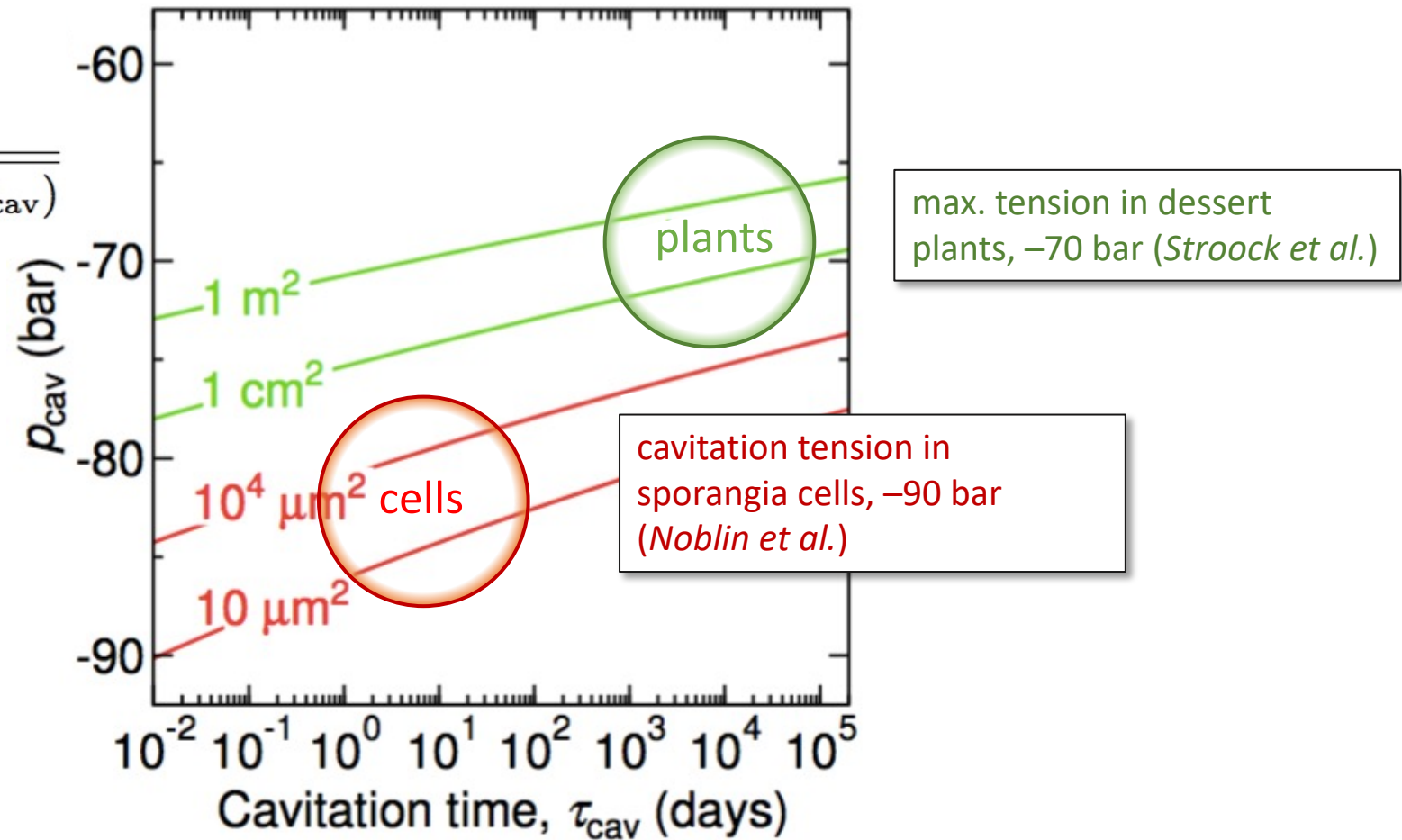
$$p_{\text{cav}} = -\frac{2}{\alpha_{\text{lip}}} \left(\frac{w_{\text{lip}}}{3}\right)^{3/2} \frac{1}{\sqrt{k_{\text{B}}T \ln(\kappa_0^{\text{lip}} A \tau_{\text{cav}})}}$$

parameters:

$$\alpha_{\text{lip}} = 0.11$$

$$\kappa_0^{\text{lip}} = 65 \pm 4 \text{ ns}^{-1} \text{ nm}^{-2}$$

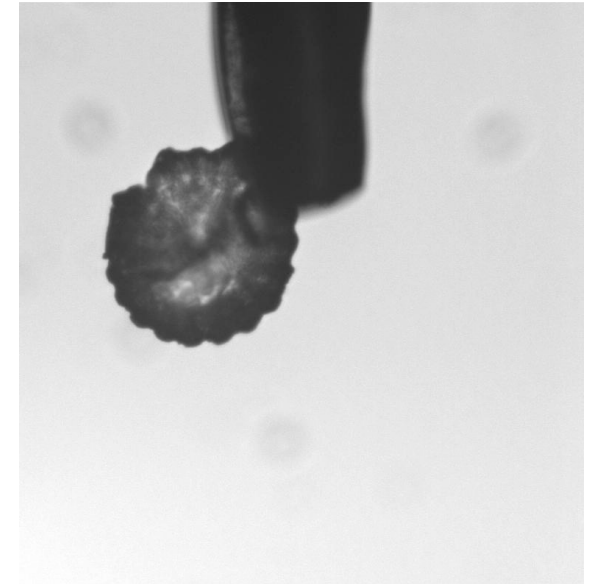
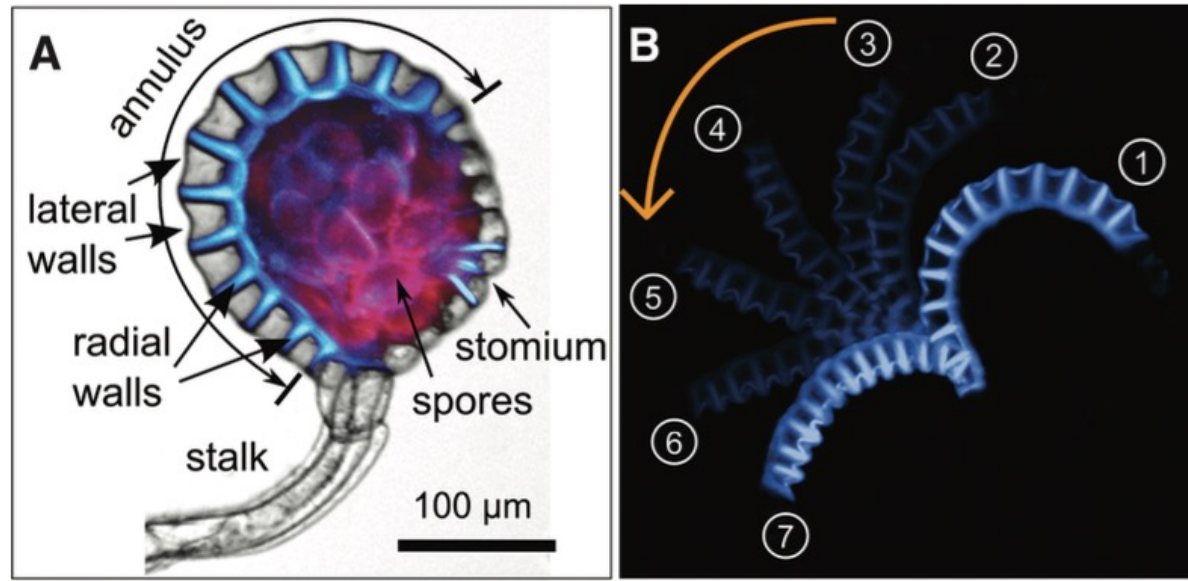
$$w_{\text{lip}} = 7.6 \text{ kJ/mol/nm}^2$$



Kanduč et al., PNAS **117**, 10733 (2020)

Water under negative pressures in cells

Catapulting mechanism of fern spores



- building tension through evaporation
- cavitation at $p_{\text{cav}} = -90$ bar

Noblin et al., Science 2012

Conclusions

- **Combining simulation and kinetic modeling approaches:**

All-atom simulations provide necessary molecular parameters

Analytic continuum model enables the extrapolation to relevant time and length scales

- **bulk water** does not likely cavitate (in homogeneous way) under negative pressures (e.g., -100 bar)

- **bilayers** can cavitate (depends on the exposed negative pressure and detailed interactions)

The presence of bilayers imposes strict upper limit of negative pressure (-70 to -90 bar) in a system

Acknowledgements

Physics

- Roland R. Netz, Free University Berlin
- Emanuel Schneck, Technical University Darmstadt
- Philip Loche, Free University Berlin

Botany

- Jochen Schenk, California State University, Fullerton
- Steven Jansen, Ulm University